

## Novel Synthesis of the 7-oxo-1,3-diazabicyclo[3.2.0]heptane and 8-oxo-1,3-diazabicyclo[4.2.0]octane Ring Systems

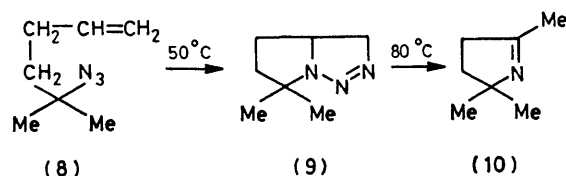
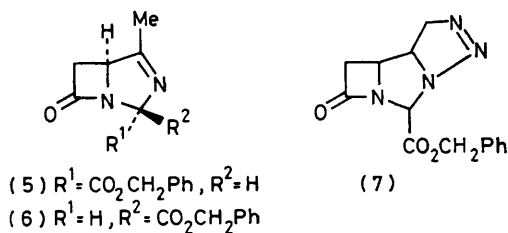
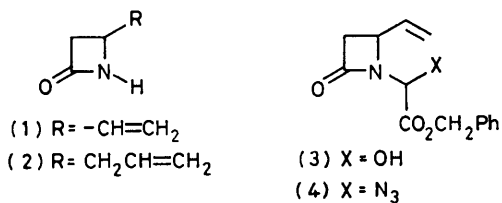
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**Summary** 4-Vinyl- and 4-allyl-azetidin-2-ones (**1**) and (**2**) have been converted into the imines (**5**) and (**19**) respectively, while use of azetidin-2-ones in which the double bond was substituted with a methoxycarbonyl group allowed the synthesis of the enamines (**13**) and (**21**).

As a continuation of our studies<sup>1</sup> concerned with the preparation of fused  $\beta$ -lactams by intramolecular cycloaddition reactions, we now report the synthesis of the 7-oxo-1,3-diazabicyclo[3.2.0]hept-3-ene and -heptane ring systems. The homologous 4,6-ring systems have also been obtained. All compounds in the following report are ( $\pm$ ) mixtures, but in some instances only one enantiomer is depicted for convenience.

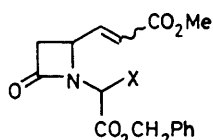
Condensation of 4-vinylazetidin-2-one (**1**)<sup>2</sup> with benzyl glyoxylate afforded the  $\alpha$ -hydroxy-ester (**3**)<sup>†</sup> which was converted into the azide (**4**)<sup>†</sup> by established procedures.<sup>3</sup> When (**4**) was heated in refluxing toluene (1 mg ml<sup>-1</sup>, under argon), the two imines (**5**)<sup>†</sup> (35%),  $\delta$  5.95 (2-H) and (**6**)<sup>†</sup> (23%), m.p. 76–77 °C,  $\delta$  5.23 (2-H) were isolated.<sup>‡</sup> The intermediate 1,2,3-triazoline (**7**) is assumed to undergo ready loss of nitrogen to form the isolated products. Heating<sup>4</sup> (**8**) has been shown to give (**10**), the reaction proceeding *via* the isolable triazoline (**9**).



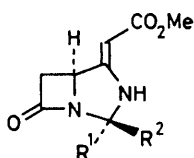
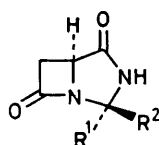
<sup>†</sup> Satisfactory elemental analysis and/or accurate mass data were obtained.

<sup>‡</sup> In all cycloadditions described, both C(2) epimers are formed. However, epimerisation to the epimer possessing the natural penicillin stereochemistry at C(2) was achieved by treatment of the mixture with 1,5-diazabicyclo[5.4.0]undec-5-ene (DBU) in methylene dichloride at -20 °C.

Ozonolysis of (3), followed by addition of methoxy-carbonylmethylenetriphenylphosphorane provided the  $\alpha\beta$ -unsaturated ester (11)<sup>†</sup> as an inseparable mixture of *E*- and *Z*-isomers. The azide (12)<sup>†</sup> was then prepared and heated in toluene at 110 °C for 23 h to give a mixture of enamines (13)<sup>†</sup> (24%, m.p. 154.5–156 °C,  $\delta$  5.54 (2-H) and (14)<sup>†</sup> (15%), m.p. 137–139.5 °C,  $\delta$  5.02 (2-H)). The olefinic proton of each epimer appeared at  $\delta$  ca. 4.7, a chemical shift consistent with an enamine structure. Although the azido-olefin (12) was a mixture of geometric isomers, each enamine epimer was a single olefinic isomer. An *X*-ray study of the epimer with the natural penicillin stereochemistry at C(2) showed that the double bond had the *Z*-configuration. Presumably the *Z*-olefin is favoured owing to stabilisation *via* hydrogen bonding between the enamine N–H and the carbonyl of the methyl ester. Ozonolysis of (13) and (14) in ethyl acetate at –76 °C provided (15),<sup>†</sup> m.p. 147.5–148.5 °C,  $\delta$  5.42 (2-H) and (16),<sup>†</sup> m.p. 154.5–155.5 °C,  $\delta$  5.07 (2-H), respectively.

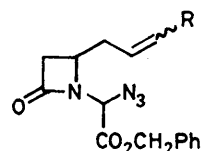


(11) X = OH

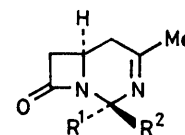
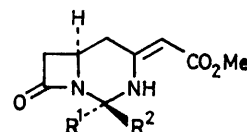
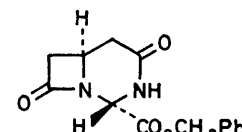
(12) X = N<sub>3</sub>(13) R<sup>1</sup> = CO<sub>2</sub>CH<sub>2</sub>Ph, R<sup>2</sup> = H(14) R<sup>1</sup> = H, R<sup>2</sup> = CO<sub>2</sub>CH<sub>2</sub>Ph(15) R<sup>1</sup> = CO<sub>2</sub>CH<sub>2</sub>Ph, R<sup>2</sup> = H(16) R<sup>1</sup> = H, R<sup>2</sup> = CO<sub>2</sub>CH<sub>2</sub>Ph

We have also demonstrated the cyclisation process in the homologous series. Thus 4-allyl-azetidin-2-one (2)<sup>5</sup> was converted into the azide (17),<sup>§</sup> which was heated in toluene at reflux for 7 h. Treatment of the crude product<sup>¶</sup> with

DBU afforded the imine (19)<sup>†</sup> (30%), m.p. 94.5–95 °C,  $\delta$ (CDCl<sub>3</sub>) 2.03 (s, CH<sub>3</sub>) and 5.72 (br. s, 2-H), possessing the natural penicillin stereochemistry at C(2).



(17) R = H

(18) R = CO<sub>2</sub>Me(19) R<sup>1</sup> = CO<sub>2</sub>CH<sub>2</sub>Ph, R<sup>2</sup> = H(20) R<sup>1</sup> = H, R<sup>2</sup> = CO<sub>2</sub>CH<sub>2</sub>Ph(21) R<sup>1</sup> = CO<sub>2</sub>CH<sub>2</sub>Ph, R<sup>2</sup> = H(22) R<sup>1</sup> = H, R<sup>2</sup> = CO<sub>2</sub>CH<sub>2</sub>Ph

(23)

The azide (18)<sup>§</sup> was also prepared and cyclised (refluxing toluene, 5½ h) to give a mixture of enamines (21) and (22), from which (21)<sup>†</sup> (20%), m.p. 149–150 °C,  $\delta$ (CDCl<sub>3</sub>) 4.66 (br. s, =CH) and 5.53 (d, *J* 1.7 Hz, 2-H),  $\lambda_{\max}$ (EtOH) 283 nm ( $\epsilon$  18,700), could be crystallised. Purification of the mother liquors on 'Florisil' gave a further quantity of material (30%) which was an inseparable mixture of epimers (21) and (22) in a ratio of 1:4. The latter showed  $\delta$ (CDCl<sub>3</sub>) 4.68 (br. s, =CH) and 5.15 (dd, *J* 1.7 and 4 Hz, 2-H). The *Z*-geometry of the double bond in each epimer is assigned by analogy with the corresponding 5-membered ring derivatives. Ozonolysis of (21) in methylene dichloride containing methanol (ca. 2%) gave the amide (23)<sup>†</sup> (87%), m.p. 141–142 °C,  $\delta$  5.61 (2-H). All compounds showed the expected spectroscopic properties; none of the bicyclic esters or corresponding free acids showed any antibacterial activity.

We thank Professor T. J. King, University of Nottingham, for *X*-ray work.

(Received, 8th June 1981; Com. 665.)

<sup>§</sup> Azides in this series were unstable oils, and were generally used immediately after preparation.

<sup>¶</sup> In this case the epimer (20) with the proton  $\alpha$  at C(2) was unstable to chromatography and could not be selectively crystallised from the mixture.

<sup>1</sup> M. J. Pearson, *J. Chem. Soc., Perkin Trans. 1*, submitted for publication.

<sup>2</sup> T. Durst and M. J. O'Sullivan, *J. Org. Chem.*, 1970, **35**, 2043.

<sup>3</sup> M. J. Pearson, *J. Chem. Soc., Perkin Trans. 1*, 1977, 189.

<sup>4</sup> A. L. Logothetis, *J. Am. Chem. Soc.*, 1965, **87**, 749.

<sup>5</sup> A. J. G. Baxter, K. H. Dickinson, P. M. Roberts, T. C. Smale, and R. Southgate, *J. Chem. Soc., Chem. Commun.*, 1979, 236.